




**EDICT
OF
GOVERNMENT**

In order to promote public education and public safety, equal justice for all, a better informed citizenry, the rule of law, world trade and world peace, this legal document is hereby made available on a noncommercial basis, as it is the right of all humans to know and speak the laws that govern them.

EAS 81-4 (2006) (English): Milk powders
Determination of titratable acidity
(Reference method)

ISO INSIDE



BLANK PAGE





EAS 81-4:2006
ICS 67.100.10

EAST AFRICAN STANDARD

Milk powders — Methods for the analysis — Part 4: Determination of titratable acidity (Reference method)

EAST AFRICAN COMMUNITY

EAS 81-4:2006

Foreword

Development of the East African Standards has been necessitated by the need for harmonizing requirements governing quality of products and services in East Africa. It is envisaged that through harmonized standardization, trade barriers which are encountered when goods and services are exchanged within the Community will be removed.

In order to achieve this objective, the Partner States in the Community through their National Bureaux of Standards, have established an East African Standards Committee.

The Committee is composed of representatives of the National Standards Bodies in Partner States, together with the representatives from the private sectors and consumer organizations. Draft East African Standards are circulated to stakeholders through the National Standards Bodies in the Partner States. The comments received are discussed and incorporated before finalization of standards, in accordance with the procedures of the Community.

East African Standards are subject to review, to keep pace with technological advances. Users of the East African Standards are therefore expected to ensure that they always have the latest versions of the standards they are implementing.

© East African Community 2006 — All rights reserved*

East African Community

P O Box 1096

Arusha

Tanzania

Tel: 255 27 2504253/8

Fax: 255-27-2504481/2504255

E-Mail: eac@eachq.org

Web: www.each.org

* © 2006 EAC — All rights of exploitation in any form and by any means reserved worldwide for EAC Partner States' NSBs.

International Standard



6091

INTERNATIONAL ORGANIZATION FOR STANDARDIZATION • МЕЖДУНАРОДНАЯ ОРГАНИЗАЦИЯ ПО СТАНДАРТИЗАЦИИ • ORGANISATION INTERNATIONALE DE NORMALISATION

Dried milk — Determination of titratable acidity (Reference method)

Lait sec — Détermination de l'acidité titrable (Méthode de référence)

First edition — 1980-10-15

UDC 637.143 : 543.241.5

Ref. No. ISO 6091-1980 (E)

Descriptors : dairy products, milk, dried foods, dried milk, chemical analysis, titration, acidity.

Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards institutes (ISO member bodies). The work of developing International Standards is carried out through ISO technical committees. Every member body interested in a subject for which a technical committee has been set up has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work.

Draft International Standards adopted by the technical committees are circulated to the member bodies for approval before their acceptance as International Standards by the ISO Council.

International Standard ISO 6091 was developed by Technical Committee ISO/TC 34, *Agricultural food products*, and was circulated to the member bodies in November 1978.

It has been approved by the member bodies of the following countries :

Australia	Hungary	Portugal
Belgium	India	Romania
Brazil	Ireland	South Africa, Rep. of
Bulgaria	Israel	Sri Lanka
Canada	Kenya	Thailand
Cyprus	Korea, Rep. of	Turkey
Czechoslovakia	Malaysia	United Kingdom
Egypt, Arab Rep. of	Netherlands	USSR
Ethiopia	New Zealand	Yugoslavia
France	Peru	
Germany, F. R.	Poland	

No member body expressed disapproval of the document.

NOTE — The method specified in this International Standard has been developed jointly with the IDF (International Dairy Federation) and AOAC (Association of Official Analytical Chemists, USA); it will also be included in the FAO/WHO Code of Principles concerning milk and milk products and associated standards.

Dried milk — Determination of titratable acidity (Reference method)

1 Scope and field of application

This International Standard specifies a reference method for the determination of the titratable acidity of all types of dried milk.

2 References

ISO/R 707, *Milk and milk products — Sampling*.

ISO/R 1736, *Dried milk — Determination of fat content (Reference method)*.

3 Definition

titratable acidity of dried milk : The number of millilitres of 0,1 mol/l sodium hydroxide solution required to titrate a quantity of the reconstituted milk corresponding to 10 g of solids-not-fat to the pH of 8,40.

4 Principle

Preparation of reconstituted milk by addition of water to a test portion of dried milk corresponding accurately to 5 g of solids-not-fat. Titration with 0,1 mol/l sodium hydroxide solution to the pH of 8,40. Multiplication of the number of millilitres used in the titration by the factor 2, in order to obtain the number of millilitres in terms of 10 g of solids-not-fat.

The amount of sodium hydroxide solution required is a function of the amount of natural buffering substances present in the product, and of developed or added acid or alkaline substances.

5 Reagent and material

All reagents shall be of recognized analytical quality. Water shall be distilled or deionized water, freed from carbon dioxide by boiling for 10 min before use.

5.1 Sodium hydroxide, standard volumetric solution, $c(\text{NaOH}) = 0,1 \pm 0,000 2 \text{ mol/l}^{1)}$, carbonate free.

Protect this solution against absorption of carbon dioxide.

5.2 Nitrogen.

6 Apparatus

6.1 Analytical balance.

6.2 pH meter, with slope control, capable of being read to 0,01 pH unit, with a glass measuring electrode and a suitable reference electrode, calibrated using two buffer solutions of pH approximately 7 and 9 respectively, known to within $\pm 0,01$ pH unit.

6.3 Magnetic stirrer.

6.4 Burette, graduated in 0,1 ml and with an accuracy of 0,05 ml.

6.5 Measuring cylinder, of capacity 50 ml.

6.6 Conical flask, of capacity 100 ml or 150 ml, with a ground neck and ground glass stopper. The neck shall be sufficiently wide to accommodate the two electrodes, the burette tip and the nitrogen line.

7 Sampling

See ISO/R 707.

8 Procedure

8.1 Preparation of the test sample

Transfer the sample to a clean, dry container (provided with an air-tight lid) of a capacity about twice the volume of the sample.

Close the container immediately and thoroughly mix the contents by repeatedly shaking and inverting the container. During these operations, exposure of the sample to the atmosphere should be avoided as far as possible, to minimize absorption of water.

1) Hitherto expressed as "0,1 \pm 0,000 2 N standard volumetric solution".

8.2 Test portion

Weigh $(500/a) \pm 0,01$ g of the test sample(8.1) into the conical flask (6.6), a being the solids-not-fat content of the sample, expressed as a percentage by mass.

NOTE — The solids-not-fat content of the sample may be calculated by subtracting the fat content (determined in accordance with ISO/R 1736) and the moisture content¹⁾ from 100.

8.3 Determination

8.3.1 Prepare reconstituted milk by adding 50 ml of water at about 20 °C to the test portion (8.2), and agitating vigorously. Allow to stand for about 20 min.

8.3.2 Titrate the contents of the conical flask by adding the sodium hydroxide solution (5.1) from the burette (6.4) until the pH, measured by means of the pH meter (6.2), persists for about 5 s at 8,40; during the titration, the solution should be stirred using the magnetic stirrer (6.3), and absorption of carbon dioxide from the air should be avoided by flushing the conical flask with the nitrogen (5.2). The titration should be completed within 1 min.

Record the volume, in millilitres, of sodium hydroxide solution used, to the nearest 0,05 ml.

9 Expression of results

9.1 Method of calculation and formula

The titratable acidity is equal to

$$2 \times V$$

where V is the volume, in millilitres, of the sodium hydroxide solution (5.1) used for the titration (8.3.2).

Express the result to one decimal place.

9.2 Repeatability

The difference between the results of two determinations carried out simultaneously or in rapid succession by the same analyst shall not exceed 0,4 ml of 0,1 mol/l sodium hydroxide solution per 10 g of solids-not-fat.

10 Test report

The test report shall show the method used and the result obtained. It shall also mention any operating conditions not specified in this International Standard, or regarded as optional, as well as any circumstances that may have influenced the result.

The report shall include all details required for the complete identification of the sample.

1) A method for the determination of moisture content is specified in IDF Standard No. 26, *Determination of the water content of dried milk*, which is currently under revision by a joint ISO/IDF/AOAC group of experts.

